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Total synthesis of novel D-ring-modified triptolide analogues: structure—cytotoxic activity relationship studies on the D-ring of triptolide†

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The total synthesis of a *trans*-position butenolide analogue of triptolide 3 and the semi-synthesis of analogue 4, with a furan ring, and compound 5, without a planar D-ring, are described. Studies into the antitumor activity of these compounds suggest that the five-membered unsaturated lactone ring (D-ring) of triptolide is essential to its potent anticancer activity and the C18 carbonyl group may exert an important influence on the interaction between triptolide and the target molecule(s) responsible for initiating their cytotoxic effects.

Triptervgium wilfordii Hook. f. (TWHF), commonly known as Lei Gong Teng (Thunder God Vine), has been used in traditional Chinese medicine to treat autoimmune and inflammatory diseases. such as rheumatoid arthritis, for centuries. 1-3 Triptolide (1) and triptonide (2) (Fig. 1), the major components responsible for the clinical properties of TWHF, were first isolated from TWHF extracts and characterized in 1972 as diterpenoid triepoxide lactones containing an 18 ($4\rightarrow3$) abeo-abietane skeleton.⁴ Right after their isolation, triptolide (1) and triptonide (2) were shown to possess potent antitumor, anti-inflammatory, immunosuppressive, and antifertility activities. 4-22 Compared to some conventional chemotherapy drugs, triptolide has a similar and even superior anticancer activity, especially against p53 mutated or multi-drug resistant cells.¹⁷ All of the antitumor properties mentioned above suggest that triptolide should be a promising anticancer drug. However, no systematic structure-cytotoxic activity relationship (SAR) studies have been reported.

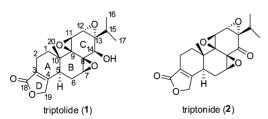


Fig. 1 Triptolide and triptonide from Tripterygium wilfordii Hook. f.

For a long time, there have been no studies on the structure-activity relationship of the D-ring of triptolide except for two

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patents,^{23,24} describing some butenolide-modified triptolide analogues without any biological activity data, and our previous paper,²⁵ reporting that an analogue (compound **6**) (Fig. 2) with a five-membered unsaturated lactam ring has the same activity as the natural triptolide. So the structure–activity relationship of the D-ring is still obscure. To explore whether the five-membered unsaturated lactone ring of triptolide is completely critical to its anticancer activity, compound **3**, having a *trans*-position butenolide, compound **4**,which has a furan ring replacing the five-membered unsaturated lactone ring, and compound **5** without the planar D-ring were synthesized for SAR studies of the D-ring. The SAR studies of these tripolide analogues were performed by using ovary (SK-OV-3) and prostate (PC-3) tumor cells

Fig. 2 Triptolide analogues 3–6.

The synthetic strategy followed for the preparation of the triptonide analogue 3 is depicted in Scheme 1. The synthesis of compound 3 started with known abietic acid 7, which was converted to C4-(18)-alkene 8 by a known procedure reported by us previously. ²⁶ Allylic oxidation of 8 with SeO₂/TBHP provided a C-3 α-alcohol 9, which was oxidized readily under Swern

Scheme 1 The total synthesis of compound 3.

conditions²⁷ to the enone **10** in 83% yield for the two steps. When compound **10** was treated with isopropoxydimethylsilylmethylmagnesium chloride in THF at $-30\,^{\circ}$ C, ²⁸ the nucleophilic addition proceeded smoothly, providing a β -hydroxysilane intermediate. This intermediate, without purification, was subjected to oxidative cleavage of the corresponding Si–C bond by potassium fluoride and 30% hydrogen peroxide to give the 1,2-diol **11** as a single stereoisomer in 80% yield. Protection of the primary hydroxyl group with acetic anhydride provided compound **12**, which was treated with SOCl₂ in Et₂O, affording **13** in 49% yield for the two steps. Ganem oxidation²⁹ of **13** with Me₃NO, including a double bond migration, gave an aldehyde intermediate, which, without purification, was treated with K₂CO₃ in MeOH and H₂O, providing hemiacetals **14** and **15** that, upon oxidation with pyridinium dichromate, delivered lactones **16** and **17** in 29% and

35% yields, respectively, for the three steps. Treatment of **16** and **17** with methoxide ion in methanol at room temperature for 10 min afforded butenolide **18**³⁰ in nearly quantitative yield. The X-ray crystallographic analysis of compound **18** confirmed its structure to be as shown in Fig. 3.³¹

With key intermediate 18 in hand, compound 18 was converted into compound 3 by a sequence similar to that developed previously for the construction of the C-ring functionality. 26,32 Treatment of 18 with ammonium ceric nitrate in H_2O-CH_3CN gave benzyl alcohol as the intermediate, which, without purification, upon oxidation of hydroxyl with pyridinium dichromate, delivered ketone 19 that was demethylated to afford phenol 20 in 85% yield for the three steps. Treatment of 20 with NaBH₄ in MeOH gave C-7 β -alcohol 21 in 84% yield, along with the C-7 α -alcohol isomer in 13% yield. Periodate oxidation³³ of 21 afforded epoxy dienone



Fig. 3 The X-ray single crystal structure of compound 18.

22. When methyl(trifluoromethyl)dioxirane generated in situ was used,³⁴ the second C-9,11 β-epoxide was introduced as a single diastereomer. Further epoxidation with basic H₂O₂ provided compound 3, with triepoxides as the sole epoxidation product, in 62% yield for the three steps. The X-ray crystallographic analysis of compound 3 confirmed its structure to be as shown in Fig. 4.31

Fig. 4 The X-ray single crystal structure of compound 3.

We then switched to the synthesis of the furan ring triptolide analogue. We used triptolide (1), which was extracted from TWHF of our region, as the starting material and initially focused our attention on modifying the butenolide on the D-ring in order to obtain the furan ring analogues (Scheme 2). Protection of the hydroxyl group with DMSO and Ac₂O in AcOH gave the methylthiomethyl ether 23 in 55% yield. Reduction of the lactone of compound 23 with DIBAL in CH₂Cl₂ at -78 °C afforded a hemiacetal intermediate, which, without purification, upon treatment with silica gel35 in CDCl3, provided the furan ring in 24 in 75% yield. Deprotection of compound 24 with HgCl₂ in CH₃CN and H₂O gave target compound 4 in 85% yield.

Scheme 2 The synthesis of analogue 4.

In order to obtain the analogues without a D-ring, we used abietic acid 7, which was converted to compound 25 by a known procedure, ²⁶ as the starting material (Scheme 3). Benzylic oxidation of 25 with Na₂Cr₂O₇ and N-hydroxyphthalimide³⁶ in acetone provided ketone 26 that was demethylated to afford the phenol 27 in 72% yield for two steps. Reduction of 27 with NaBH₄ in MeOH gave the C-7 β-alcohol 28, along with the C-7 α-alcohol isomer. Periodate oxidation³² of **28** afforded epoxy

Scheme 3 The synthesis of analogue 5.

dienone 29 in 63% yield over the two steps. Epoxidation of 29 with methyl(trifluoromethyl)dioxirane³⁴ followed by treatment with basic H₂O₂ provided compound 5, a triepoxide, as the sole epoxidation product, in 65% yield over two steps.

As shown above, we obtained four novel D-ring modified triptolide analogues with a trans-butenolide, with a furan ring, without a D-ring and with a five-membered unsaturated lactam ring and we evaluated the in vitro anticancer effects of those target compounds (3-6) against two human tumor cell lines derived from ovary (SK-OV-3) and prostate (PC-3) using sulforhodamine B (SRB) assays.37 The results revealed that analogue 6,25 with a five-membered unsaturated lactam ring, which behaves as an ester isostere, had the same effect against the two cell lines as the natural triptonide (Table 1). However, compound 3 with a transbutenolide was shown to be only weakly cytotoxic and analogue 4 with the furan ring and analogue 5 without the D-ring completely lost their cytotoxicity against PC-3 and SK-OV-3. Based on the above results, we presumed that the five-membered unsaturated lactone ring (D-ring) of triptolide is essential to its potent anticancer activity and that the C18 carbonyl group may exert an important influence on the interaction between triptolide and the target molecule(s) responsible for initiating their cytotoxic effects. However, the exact mechanism is still unclear as compounds 1, 2 and 6, having five-membered unsaturated lactone or lactam rings, are significantly cytotoxic while the furan ring analogue 4, without the C18 carbonyl group, and analogue 5, without the planar D-ring, nearly lost their cytotoxicity completely. Also, the cytotoxicity against the two cell lines of the trans-butenolide analogue 3 with the C19 carbonyl group was greatly reduced.

Table 1 The in vitro anticancer activity of the triptolide analogues in SK-OV-3 and PC-3 cells

Compound	$IC_{50}(\mu M)$	
	SK-OV-3	PC-3
1	0.006	0.02
2	0.008	0.6
3	6.5	> 100
4	> 100	> 100
5	> 100	> 100
6	0.02	1.4

In summary, four novel D-ring-modified triptolide analogues (3–6) with a *trans*-butenolide, a furan ring, without a D-ring and with a five-membered unsaturated lactam ring were synthesized and tested for their cytotoxicity against two human tumor cell lines. The effect of the five-membered unsaturated lactone ring (D-ring) on triptolide's potent anticancer activity has been unclear for a long time. From the current investigation, the structure–activity relationships of these compounds suggest that the five-membered unsaturated lactone ring (D-ring) of triptolide is essential to its potent anticancer activity and that the C18 carbonyl group may exert an important influence on the interaction between triptolide and the target molecule(s) responsible for initiating their cytotoxic effects. More importantly, it will give an impetus to the systematic structure–cytotoxic activity relationship studies of triptolide.

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